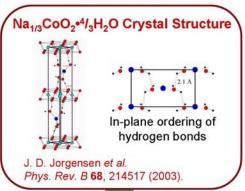
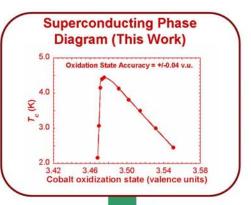
## Superconductivity and Cobalt Oxidation State in $Na_xCoO_2$ •4xH<sub>2</sub>O (x ~ $\frac{1}{3}$ )

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**Motivation:** To investigate the effects of synthetic chemistry, temperature and pressure on the oxidation state of cobalt in superconducting Na<sub>x</sub>CoO<sub>2</sub>·4xD<sub>2</sub>O ( $x \sim \frac{1}{3}$ ).



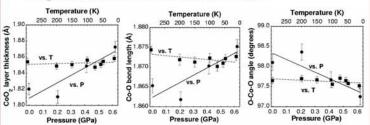
## Previously Reported Superconducting Phase Diagrams hole doped w.r.t. Co<sup>3+</sup> electron doped w.r.t. Co<sup>4+</sup> 5.0 Miline et al. 3.5 3.0 Co<sup>3+</sup> 2.5 3.0 Co<sup>3+</sup> 2.5 2.0 3.0 3.2 3.4 3.6 3.8 4.0



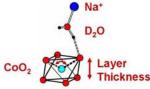


## Temperature and Pressure Effects on the Crystal Structure of Na<sub>1/3</sub>CoO<sub>2</sub>•4/<sub>3</sub>D<sub>2</sub>O

 Thermal expansion and compressibility were monitored using neutron powder diffraction. As expected, anisotropic thermal expansion occurs (Δc/c > Δa/a) with increasing T; however, the CoO<sub>2</sub> layer thickness increases with increasing P.



 CoO<sub>2</sub> layer thickness is controlled by the Co–O bond length and the O–Co–O angle.

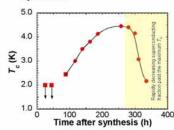


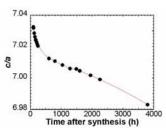
 This increase in layer thickness is attributed to P-induced strengthening of the H-bonds between the CoO<sub>2</sub> layers and the D<sub>2</sub>O molecules. The strengthened H-bonds requires a charge redistribution that weakens the Co–O bonds, thus changing the electronic structure of the material (decreased Co valence).

J. D. Jorgensen et al., Phys. Rev. B 72, 224515 (2005).

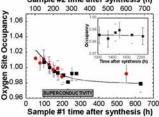
## Determination of the Superconducting Phase Diagram Using a Single Sample

T<sub>c</sub> and the c/a ratio of metastable Na<sub>1/3</sub>CoO<sub>2</sub>·4/<sub>3</sub>D<sub>2</sub>O (made in one step using a Br<sub>2</sub>/D<sub>2</sub>O solution) vary as a function of time after synthesis.





• These changes are due to oxygen vacancy formation within the CoO<sub>2</sub> planes. Sample #2 time after synthesis (h)



At high Co oxidation states,  $Na_{1/3}CoO_2^{-4}I_3D_2O$  is metastable. Oxygen vacancies form in the  $CoO_2$  planes, lowering  $Co^{n+}$ .  $T_c$  increases as  $Co^{n+}$  decreases, reaching a maximum near  $n \approx +3.5$ , followed by a rapid decrease attributed to charge ordering.

**Impact:** This work illustrates how neutron powder diffraction is combined with other techniques to probe important structural variables such as ordering, defect concentrations and charge states that impact a material's physical properties.

Future Work: This project is part of an ongoing program on layered superconductors. Future work will focus on searching for superconductivity in new layered compounds, including structural analogs to Na<sub>x</sub>CoO<sub>2</sub>•4xH<sub>2</sub>O, compounds with the K<sub>2</sub>NiF<sub>4</sub> structure, and layered nitrides, phosphides, and borides.

P. W. Barnes, M. Avdeev, J. D. Jorgensen, D. G. Hinks, H. Claus, and S. Short. Phys. Rev. B 72, 134515 (2005).







